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U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

ATTORNEY'S DOCKET NUMBER

**TRANSMITTAL LETTER TO THE UNITED STATES
DESIGNATED/ELECTED OFFICE (DO/EO/US)
CONCERNING A FILING UNDER 35 U.S.C. 371**

198928US0PCT

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR

09/700684

INTERNATIONAL APPLICATION NO.
PCT/FR00/00675INTERNATIONAL FILING DATE
17 March 2000PRIORITY DATE CLAIMED
19 March 1999

TITLE OF INVENTION

**ELECTROCHEMICAL DEVICE OF ELECTRICALLY CONTROLLABLE SYSTEM TYPE HAVING VARIABLE
OPTICAL AND/OR ENERGY PROPERTIES**

APPLICANT(S) FOR DO/EO/US

Claude MORIN, et al.

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. ☒ This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. ☒ This is an express request to begin national examination procedures (35 U.S.C. 371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. 371(b) and PCT Articles 22 and 39(1).
4. ☐ A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date.
5. ☒ A copy of the International Application as filed (35 U.S.C. 371 (c) (2))
 - a. ☐ is transmitted herewith (required only if not transmitted by the International Bureau).
 - b. ☒ has been transmitted by the International Bureau.
 - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US).
6. ☒ A translation of the International Application into English (35 U.S.C. 371(c)(2)).
7. ☒ A copy of the International Search Report (PCT/ISA/210).
8. ☒ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
 - a. ☐ are transmitted herewith (required only if not transmitted by the International Bureau).
 - b. ☐ have been transmitted by the International Bureau.
 - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
 - d. ☒ have not been made and will not be made.
9. ☐ A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
10. ☐ An oath or declaration of the inventor(s) (35 U.S.C. 371 (c)(4)).
11. ☐ A copy of the International Preliminary Examination Report (PCT/IPEA/409).
12. ☐ A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).

Items 13 to 18 below concern document(s) or information included:

13. ☐ An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
14. ☐ An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
15. ☒ A **FIRST** preliminary amendment.
A **SECOND** or **SUBSEQUENT** preliminary amendment.
16. ☐ A substitute specification.
17. ☐ A change of power of attorney and/or address letter.
18. ☐ Certificate of Mailing by Express Mail
19. ☒ Other items or information:

Request for Consideration of Documents Cited in International Search Report

Notice of Priority

PCT/IB/304

PCT/IB/308

Drawings (4 Sheets)

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR 1.53) 09/700684	INTERNATIONAL APPLICATION NO. PCT/FR00/00675	ATTORNEY'S DOCKET NUMBER 198928US0PCT
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20. The following fees are submitted:

BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)) :

- ☒ Search Report has been prepared by the EPO or JPO \$860.00
- ☐ International preliminary examination fee paid to USPTO (37 CFR 1.482) \$690.00
- ☐ No international preliminary examination fee paid to USPTO (37 CFR 1.482) but international search fee paid to USPTO (37 CFR 1.445(a)(2)) \$710.00
- ☐ Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO \$1000.00
- ☐ International preliminary examination fee paid to USPTO (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(2)-(4) \$100.00

ENTER APPROPRIATE BASIC FEE AMOUNT =**CALCULATIONS PTO USE ONLY**

Surcharge of \$130.00 for furnishing the oath or declaration later than months from the earliest claimed priority date (37 CFR 1.492 (e)).

☐ 20 ☒ 30

\$860.00

\$130.00

CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE		
Total claims	37 - 20 =	17	x \$18.00		\$306.00
Independent claims	1 - 3 =	0	x \$80.00		\$0.00

Multiple Dependent Claims (check if applicable).

☐

\$0.00

TOTAL OF ABOVE CALCULATIONS =

\$1,296.00

Reduction of 1/2 for filing by small entity, if applicable. Verified Small Entity Statement must also be filed (Note 37 CFR 1.9, 1.27, 1.28) (check if applicable).

☐

\$0.00

SUBTOTAL =

\$1,296.00

Processing fee of \$130.00 for furnishing the English translation later than months from the earliest claimed priority date (37 CFR 1.492 (f)).

☐ 20 ☐ 30

+

\$0.00

TOTAL NATIONAL FEE =

\$1,296.00

Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31) (check if applicable).

☐

\$0.00

TOTAL FEES ENCLOSED =

\$1,296.00

Amount to be:

refunded

\$

charged

\$

☒ A check in the amount of \$1,296.00 to cover the above fees is enclosed.

☐ Please charge my Deposit Account No. _____ in the amount of _____ to cover the above fees.
A duplicate copy of this sheet is enclosed.

☒ The Commissioner is hereby authorized to charge any fees which may be required, or credit any overpayment to Deposit Account No. 15-0030 A duplicate copy of this sheet is enclosed.

NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO:

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REGISTRATION NUMBER

DATE

11-17-00

09/17/00 09/17/00 684
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198928US0PCT

IN THE UNITED STATES PATENT & TRADEMARK OFFICE

IN RE APPLICATION OF: :
CLAUDE MORIN ET AL : ATTN: APPLICATION DIVISION
SERIAL NO: NEW APPLICATION :
(BASED ON PCT/FR00/00675)
FILED: HEREWITH :
FOR: ELECTROCHEMICAL DEVICE OF
ELECTRICALLY CONTROLLABLE
SYSTEM TYPE HAVING VARIABLE
OPTICAL AND/OR ENERGY
PROPERTIES

PRELIMINARY AMENDMENT

ASSISTANT COMMISSIONER FOR PATENTS
WASHINGTON, D.C. 20231

SIR:

Prior to examination on the merits, please amend the above-identified application as follows:

Page 1, before line 1, delete the title of the invention in its entirety, and insert therefor:

--ELECTROCHEMICAL DEVICE, SUCH AS AN ELECTRICALLY
CONTROLLED SYSTEM WITH VARIABLE OPTICAL AND/OR ENERGY
PROPERTIES - -.

IN THE CLAIMS

Please cancel Claims 1-21 and replace with the following claims:

--22. An electrochemical device comprising at least one carrier substrate provided with a stack of functional layers comprising at least one electrically conducting layer A based on metal oxide(s) and at least one electrochemically active layer F, wherein said layer A is part of a multicomponent electrode E combining with the layer A at least one higher-conductivity material B and/or at least one network C of conducting wires or of conducting strips.

23. The device of Claim 22, which is an electrically controllable system having variable optical and/or energy properties.

24. The device of Claim 22, wherein the material B is in the form of at least one layer combined with the layer A and in electrical contact therewith.

25. The device of Claim 22, wherein the material B is incorporated in the layer A.

26. The device of Claim 25, wherein the the material B is incorporated in the layer A, in the form of fibers or particles.

27. The device of Claim 22, wherein the layers A are based on doped metal oxides selected from the group consisting of doped tin oxide, doped zinc oxide and doped indium oxide.

28. The device of Claim 27, wherein the doped tin oxide is fluorine doped or antimony doped.

29. The device of Claim 27, wherein the doped zinc oxide is aluminum doped, tine doped or fluorine doped.

30. The device of Claim 27, wherein the doped indium oxide is tin doped.
31. The device of Claim 22, wherein the material B is essentially metallic.
32. The device of Claim 31, wherein the material B is based on metals selected from the group consisting of Ag, Au, Cu, Al, and alloys of thereof
33. The device of Claim 22, wherein said multicomponent electrode E is essentially transparent in the visible region.
34. The device of Claim 22, wherein said network C includes a plurality of conducting strips obtained by screen printing using a paste-like suspension of silver-type metal and a low-melting-point frit in an organic binder.
35. The device of Claim 34, wherein said plurality of conducting strips are essentially parallel each to the other.
36. The device of Claim 22, wherein the network C is screen-printed onto the glass-type carrier substrate, then covered with at least one electrically conducting layer A in order to form an electrode E, or is laid down on the electrically conducting layer A covering the carrier substrate.
37. The device of Claim 22, wherein the network C includes a plurality of conducting wires in the form of essentially metallic wires surface-laid on a sheet based on thermoplastic polymer.
38. The device of Claim 22, wherein the network C is based on a fabric, a net or a metallic nonwoven material.
39. The device of Claim 38, wherein the network C is comprises metallic wires having a diameter of from 10 to 100 μm .

40. The device of Claim 38, wherein the network C is a network surface-laid on a sheet based on thermoplastic polymer.

41. The device of Claim 22, wherein the network C is obtained by etching or perforating a metallic layer or a metallic sheet.

42. The device of Claim 22, wherein the multicomponent electrode E comprises at least one layer A and at least one layer B in electrical contact, at least one of these layers optionally being in contact with at least one layer D of dielectric material, and all of the layers A, B and D preferably forming a stack of layers with interference interaction.

43. The device of Claim 22, wherein the layers D have an optical function, a function of anchoring the other layers B to the carrier substrate or a function as a barrier to the migration of alkaline species coming from the glass.

44. The device of Claim 43, wherein the layers D are in the form selected from the group consisting of metal oxide, silicon oxide, metal oxycarbide, silicon oxycarbide, metal oxynitride, silicon oxycarbide and silicon nitride.

45. The device of Claim 22, wherein the multicomponent electrodes E comprise the sequence ITO/Ag/ITO or Ag/ITO with optional interposition of thin layers of partially oxidized metal at the Ag/ITO interface.

46. The device of Claim 22, wherein the multicomponent electrodes E are provided with current leads.

47. The device of Claim 46, wherein the current leads are in the form of metal braids or shims.

48. The device of Claim 22, which is an electrochromic system, with at least one carrier substrate and a stack of functional layers comprising at least, in succession, a first electrically conducting layer, an electrochemically active layer liable to reversible insertion of ions of anodic-coloring or, respectively, cathodic-coloring electrochromic material type, a layer of electrolyte, a second electrochemically active layer liable to reversible insertion of ions of anodic-coloring or, respectively, anodic-coloring electrochromic material type, and a second electrically conducting layer, with at least one of the two electrically conducting layers in the form of a layer A based on metal oxide(s) and being part of a multicomponent electrode E.

49. The device of Claim 48, wherein said electrochromic system is an all-solid or all-polymer electrochromic system.

50. The device of Claim 48, wherein said ions are H^+ , Li^+ or OH^- , of cathodic-coloring

51. The device of Claim 22, which is a viologenic system with at least one carrier substrate and a stack of functional layers comprising at least, in succession, a first electrically conducting layer, a film having viologenic properties in the form of a polymer, of a gel or of a suspension in a liquid medium, and a second electrically conducting layer, with at least one of the two electrically conducting layers being of type A based on metal oxide(s) and being part of a multicomponent electrode E.

52. The device of Claim 22, wherein the stack of functional layers is arranged between two substrates, each of which may be rigid, of glass type or rigid polymer or semi-rigid or flexible of PET type.

53. The device of Claim 52, wherein said polymer is a polycarbonate or PMMA.

54. The device of Claim 52, wherein , the layers are transparent or absorbent.

55. A glazing comprising the device of Claim 22, wherein said device using as carrier substrate at least one of the rigid constituent substrates of the glazing and/or at least one flexible substrate combined by lamination with one of the rigid constituent substrates of said glazing.

56. A method of making a glazing comprising laminating the device of Claim 22 with at least one rigid constituent substrate.

57 A electrochemical energy storage device comprising the device of Claim 22.

58. The electrochemical energy storage device of Claim 57 which is a battery or fuel cell.--

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REMARKS

Claims 22-58 are active in the present application. The claims are amended for clarity and to remove multiple dependencies. Support for Claim 22-59 is found in Claim 1-21 and the specification as originally filed. No new matter is added. An action on the merits and allowance of the claims is solicited.

Respectfully submitted,

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**ELECTROCHEMICAL DEVICE OF ELECTRICALLY CONTROLLABLE
SYSTEM TYPE HAVING VARIABLE OPTICAL AND/OR ENERGY
PROPERTIES**

5 The invention relates to electrochemical
devices, in particular of the type comprising at least
one carrier substrate provided with a stack of
functional layers comprising at least one electrically
conducting layer and at least one electrochemically
10 active layer. More particularly, the invention
envisages electrically controllable systems having
variable optical and/or energy properties, in glazing
applications or mirror applications.

 The reason for this is the increasing demand
15 for so-called "smart" glazing whose properties are
capable of variation.

 Thus, from the thermal standpoint, glazing
whose transmission/absorption can be varied within at
least part of the solar spectrum allows the solar heat
20 influx into rooms or passenger areas/compartments to be
controlled when it is fitted as external glazing in
buildings or as windows in means of transportation of
the type comprising cars, trains, aircraft, etc., thus
allowing avoidance of excessive heating of the latter
25 in strong sunlight.

 From the optical standpoint, the glazing allows
the visibility to be controlled, and when it is
installed as exterior glazing this makes it possible to
prevent glare in strong sunlight. It may also have a
30 particularly advantageous shutter effect, either used
as exterior glazing or if used as interior glazing, for
example for equipping internal partitions between rooms
(offices in a building), or for isolating compartments
in trains or aircraft.

35 There are many other applications: for example,
glazing having variable light transmission/reflection
may be used to make rear-view mirrors, which can darken
as required in order to prevent dazzling of the driver
of the car. They may also be used for road sign panels,

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or for any display panel, for example so as to reveal the pattern or message only intermittently, in order to attract greater attention.

One particularly interesting application of systems having variable light absorption relates to display screens, in particular those with which televisions and computing hardware are equipped. The reason for this is that this type of glazing makes it possible to improve the contrast of the image, especially taking ambient brightness into account.

The interest which glazing of this type can arouse is the reason for the many systems which have already been studied.

Two systems are of more particular interest for the invention: viologenic systems and electrochromic systems.

Viologenic systems allow the transmission or absorption of glazing which incorporates them to be modified, essentially in the visible region. They generally comprise just one "active layer" based on polymer, on gel or on liquid comprising a so-called cathodic active material, such as viologenic molecules, together with a so-called anodic active material, such as dimethylferrocene or phenazines. Some examples of these are described in patents EP-0 612 826 and US-5 239 406.

As is known, electrochromic systems include a layer of an electrochromic material capable of reversible and simultaneous insertion of ions and electrons and whose oxidation states corresponding to the inserted and ejected states have a distinct colour, one of the states having higher light transmission than the other, the insertion or ejection reaction being controlled by a suitable electrical supply. The electrochromic material, usually based on tungsten oxide, must therefore be placed in contact with a source of electrons, for example a transparent electrically conducting layer, and with a source of

ions (cations or anions), for example an ionic conductive electrolyte.

It is moreover known that, in order to secure at least about a hundred switching operations, there must be associated with the layer of electrochromic material a counterelectrode which is itself capable of reversibly inserting cations, symmetrically with respect to the layer of electrochromic material so that, macroscopically, the electrolyte appears as a single ion medium.

The counterelectrode must consist of a layer which is either neutral in colour or at least transparent or with little colour when the electrochromic layer is in the coloured state. As tungsten oxide is a cathodic electrochromic material, that is to say a material whose coloured state corresponds to the most reduced state, the material used for the counterelectrode is generally an anodic electrochromic material based on nickel oxide or on iridium oxide. It has also been proposed that the material used be optically neutral in the oxidation states concerned, for example cerium oxide or organic materials such as electronically conductive polymers (polyaniline, etc.) or Prussian blue.

A description of systems of this type is found, for example, in European Patents EP-0 338 876, EP-0 408 427, EP-0 575 207 and EP-0 628 849.

Currently, these systems may be arranged in two categories, according to the type of electrolyte which they use:

→ either the electrolyte is in the form of a polymer or of a gel, for example a polymer exhibiting protonic conduction, such as those described in European Patents EP-0 253 713 and EP-0 670 346, or a polymer exhibiting lithium-ion conduction, such as those described in Patents EP-0 382 623, EP-0 518 754 or EP-0 532 408,

→ or the electrolyte is an inorganic layer which is ionically conductive but electronically insulating; these systems are referred to as "all-solid" electro-

chromic systems. For the description "all-solid" applied to an electrochromic system reference may be made to the European Patent Applications EP-0 867 752 and EP-0 831 360.

5 There are other types of electrochromic systems. Mention may therefore be made of "all-polymer" electrochromic systems, where two electrically conducting layers are arranged on either side of a stack comprising a cathodic-colouring polymer, an electrically
10 insulating but ionically conductive polymer (very particularly conductive for H^+ or Li^+) and finally an anodic-colouring polymer (such as polyaniline or polypyrrole).

 Finally, there are also "active" systems in the
15 sense of the invention which combine viologenic materials and electrochromic materials, for example having the sequence conducting electrode/inorganic layer or polymer having electrochromic properties/
 layers (liquid, gel, polymer) having viologenic
20 properties/conducting electrode.

 These systems comprising reversible insertion material(s) are of particular interest in the sense that they allow modification of absorption within a wavelength region which is wider than that for
25 viologenic systems: they can absorb variably not only in the visible region but also, in particular, in the infrared region, and this can make them useful optically and/or thermally.

 A point which is common to these different
30 systems, described below by the term "active" systems is that their transmission/absorption state is controlled by applying a potential difference to their terminals, generally formed by two electrically conducting layers between which the electrochemically
35 active layer(s) lie(s). When these systems are part of "active" glazing, the electrically conducting layers are preferably transparent (or at least one of them is transparent if the other is chosen to reflect in the visible region in a mirror application). The material

needed when selecting the nature of these electrically
conducting layers has therefore to be both sufficiently
conductive and sufficiently transparent in the ranges
of thickness usually encountered in the field of thin
5 layers. The selection usually concentrates on a doped
metal oxide material, such as fluorine-doped tin oxide
($\text{SnO}_2:\text{F}$) or tin-doped indium oxide (ITO), which may be
laid down on various substrates, either hot (in
particular by pyrolysis on glass, as in the CVD method)
10 or cold (vacuum techniques of cathodic sputtering
type).

However, it has been found that, at thicknesses
at which they are still transparent, layers based on
this type of material are not fully satisfactory even
15 though they do permit the functioning of active
systems.

They are insufficiently conducting and increase
the response time of the active systems on application
to their terminals of an electrical supply appropriate
to change their transmission/adsorption state (the
20 state described below for greater simplicity by the
term "coloration" state, even if the modification of
properties also operates outside the visible region).

In addition to the fact that they reduce the
25 switching speed of the systems ("switching" or
"response time" being the period of time necessary for
the entire active system to have changed its coloration
state) the layers contribute to the creation of an edge
phenomenon, i.e. non-uniformity in the change of state
30 of the system within its surface, with a change in
"coloration" in the sense of the invention which is
almost immediate in the zones near to the current leads
supplying the electrically conducting layers, arranged
at the periphery of the systems, and which propagates
35 progressively toward the centre of the surface of the
active systems. Now, in certain applications, in parti-
cular glazing for buildings or automobiles, the final
user generally desires the fastest possible response
time and may, in addition, prefer a progressive,

uniform change of coloration over the entire surface of the active glazing.

The object of the invention is therefore to improve the performance of the electrically conducting layers of the "active" systems defined below, and very particularly of "active" glazing comprising these latter, the improvement being aimed in particular at their electrical conductivity in association with their optical properties.

A first object of the invention is an electrochemical device, in particular an electrically controllable system having variable energy and/or optical properties, including at least one carrier substrate provided with a stack of functional layers comprising at least one electrically conducting layer A based on metal oxide(s) and at least one electrochemically active layer F. The invention consists in the fact that said layer A is part of a multicomponent electrode E combining with the layer A at least one higher-conductivity material B and/or at least one network C of conducting wires or conducting strips.

For the purposes of the invention, "higher-conductivity" describes a material B which, in the form of a layer, has a surface resistance "surface R" lower than that of the layer A. For the purposes of the invention, again, "combination" indicates that the elements concerned are electrically connected one to the other, either by direct contact or else through conducting elements/layers.

The reason for this is that increasing the thickness of the layer A to heighten its conductivity (i.e. lower its surface resistance) is a limited solution: firstly in terms of cost and of manufacturing time for the layer in question and secondly in terms of optical appearance: above a certain thickness layers of this type begin to absorb in the visible region. Now, very particularly for active glazing, as required by its application, it is generally desirable to ensure maximum light transmission in the "uncoloured" state.

The solution according to the invention has therefore consisted in reconciling conductivity and transparency by developing two versions, alternative or cumulative.

5 The material B defined above may be combined with the layer A in two different ways: according to a first version it may be in the form of at least one layer combined with A and in electrical contact therewith.

10 The characteristics and thicknesses of the layers may then be optimized so that the multicomponent electrode which combines them has, overall, the required levels of transparency and surface resistance.

A second version consists in incorporating the material B in the layer A, in particular in the form of
15 fibres or of small particles. It is also possible to use a layer A based on doped oxide, for example $\text{SnO}_2 : \text{F}$, which is laid down in a known manner by liquid pyrolysis using appropriate organometallic precursors and adding, to the liquid phase comprising
20 these precursors, fibres or metallic particles, or spraying them at the surface of the substrate simultaneously with the liquid phase (for example fibres with a diameter of the order of $10 \mu\text{m}$ and a length of about 1 mm). The location of the fibres in
25 the layer is random, "percolating" over the surface of the substrate coated in this way. In this situation the doped metal oxide of the layer A also fulfils the function of fixing the metallic fibres B.

The third version consists in combining the
30 layer(s) of type A with a network of conducting elements, in particular based on a metal of intrinsically higher conductivity than the material of type A. Actually, as described in detail below, this network may be composed of linear elements visible to
35 the eye at close quarters but sufficiently discreet to be compatible with most of the applications envisaged in glazing for buildings or for vehicles. It is then advantageous for these elements to be dimensioned and arranged so that their visibility is as low as

possible. Generally, success can be obtained by making this network almost impossible to discern, at least when the system is in the coloured state.

5 It will be noted that, in the same active system, these different versions are alternative or cumulative.

10 The point common to these versions is that the additional conducting element, namely the material B or the network C, allows the overall electrode formed in this way to overcome a threshold of conductivity, so that the entirety of the electrode experiences the same potential difference at almost the same time as soon as the voltage is applied to the system, and this significantly reduces the switching time and reduces or
15 even eliminates the "coloration front" effect mentioned above. And this highly interesting technical result is not obtained by compromising the optical quality of the system if:

- 20 ➤ either this additional element itself has little or no absorbency in the visible region, so that being transparent it does not noticeably change the appearance of the glazing or the range of transmission/absorption within which variation can be made using the action of an electrical supply (type B layer),
- 25 ➤ or this additional element is sufficiently discreet not to have any adverse effect on the overall aesthetics of the active system (type C network).

The layer(s) A is/are advantageously based on metal oxide rendered conductive by doping. This may in
30 particular be doped tin oxide, in particular doped with a halogen of the type comprising fluorine ($\text{SnO}_2 : \text{F}$) or with antimony ($\text{SnO}_2 : \text{Sb}$), or zinc oxide doped, for example, with aluminium ($\text{ZnO} : \text{Al}$) or with tin ($\text{ZnO} : \text{Sn}$) or with fluorine ($\text{ZnO} : \text{F}$) or with indium ($\text{ZnO} : \text{In}$). It
35 may also be tin-doped indium oxide, such as ITO.

According to the first version the layer(s) B is/are advantageously essentially metallic, in particular based on at least one noble metal or based on an alloy comprising a noble metal of the type

comprising silver Ag or gold Au or copper Cu or aluminium Al. The layer selected is preferably one based on an alloy of silver with another metal, such as nickel or titanium. The reason for this is that it makes the layer much less susceptible to oxidation, in particular when it is in electrical contact with layers of electrochromic materials in an "all-solid" system. Gold is a material even less susceptible than pure silver to oxidation, but it is less satisfactory from an optical standpoint, being less neutral in transmission. The combination of a type A layer and a type B layer is of particular interest: as already seen, it allows sufficient enhancement of the electrical conductivity of type A layers with layers B which have low thickness and therefore no excessive adverse optical effect. This is also a novel means of incorporating layers B in electrodes, in particular silver layers, the use of which has hitherto posed the problem of protecting these from attack, in particular by oxidants. Thus type A layers can be used to "protect" type B layers, in particular from oxidation/degradation, the layers A having a double function as protection and electrical conductor, indeed a triple function with an added optical function when the thicknesses of the type A layers are adjusted as a function of those of the type B layers to optimize the optical appearance of the entirety by interference interaction between these different layers. It is thus possible, for example, to reduce the light reflection induced by the layer B.

As mentioned above, the characteristics of the layers incorporated in the multicomponent electrodes of the invention are chosen so that they are essentially transparent in the visible region.

According to the second version the small elements of the type comprising fibres, particles or grains which are incorporated in the layer may be of the metal also mentioned for the layer B: Ag, Au, Cu, Al, or else based on steel, Cr, Ni alloy, etc.

In a first embodiment of the third version of the invention, the network C advantageously includes a plurality of conducting strips, in particular essentially parallel each to the other and obtained by screen printing, using a paste-like suspension of silver-type metal and a low-melting-point frit in an organic binder. The screen printing may be carried out on a glass-type carrier substrate, which may then be covered with at least one electrically conducting layer A in order to form an electrode according to the invention. One version consists in carrying out the opposite operation, namely laying down the network C on the electrically conducting layer A. The technique of deposition by screen printing on glass is known per se for laying down conducting networks for other applications, and very particularly for forming heater networks for vehicle glazing, in order to allow demisting or defrosting by resistance heating. For further details of this technique reference may be made, for example, to the patents FR-1 464 585 and EP-0 785 700. Since the desired function here is different, the specialist will need to determine the strip width and the strip spacing which are most appropriate to give the best compromise between conductivity and aesthetics (for example a strip width of 0.1 to 0.5 mm and a strip spacing of 1 to 5 mm).

According to a second embodiment, the network C includes a plurality of essentially metallic conducting wires, preferably surface-laid on a sheet based on thermoplastic polymer. As for the first embodiment, there are known techniques for laying down conducting wires on films, for example polyvinyl butyral films, which are combined by lamination with glass to form laminated glazing, the network having a demisting/defrosting function by means of resistance heating. It is therefore possible to adapt these techniques, optionally with adjustment of the configuration, the spacing and the dimensioning of the wires, to design a network which, combined with the thermoplastic film,

will be pressed onto the type A layer, itself arranged
 on the remainder of the stack of functional layers of
 the active system, in particular by a lamination
 technique. For further details concerning the technique
 5 of laying down these wires reference may be made, in
 particular, to the patents EP-0 785 700, EP-0 553 025,
 EP-0 506 521 and EP-0 496 669. The wires may be laid
 down in the form of curves or straight lines. In
 10 outline the process consists in using a heated pressure
 roll to press the wire onto the surface of the
 thermoplastic film, the wire being fed to the pressure
 roll from a supply reel with the aid of a wire-guiding
 device.

According to another embodiment, there may be a
 15 wider interpretation of the network C, and this may in
 particular be two-dimensional, in the form of a fabric,
 net or nonwoven obtained by weaving or knitting and
 sufficiently fine and/or with a sufficiently high mesh
 size not to impair visibility. There may also be
 20 introduction of this type of material between the sheet
 based on thermoplastic polymer which serves in parti-
 cular for the lamination of the system and the type A
 layer.

This type of material, which is flexible, may
 25 preferably be obtained using metallic wires, in
 particular with a diameter between 10 μm and 100 μm .
 The mesh size, the spacing of the knit and the type of
 weaving may be modified as appropriate. Thus,
 preference is given to wires with a diameter of from 15
 30 to 25 μm with a knit structure and with mesh spacings
 of the order of from 1 to 3 mm.

The "network" also comprises metallic layers of
 sufficient thickness to give a severe reduction in
 light transmission, or even to be opaque, and which
 35 undergo treatments so that they are rendered
 discontinuous. This may be a treatment by etching a
 metallic layer laid down by cathodic sputtering, where
 the etching may be effected by laser in order to leave
 "wires" (for example of width 0.3 mm and separation of

1.5 mm) or a two-dimensional grid. The metal in the layer may be stainless steel, copper, silvered copper, aluminium, or gold in particular.

It is also possible to treat the metallic layer by perforating it to give regularly distributed openings. This metallic layer may also be replaced by a perforated metallic sheet inserted between the stack of the active system and the lamination intercalation (a sheet thicker than a layer, for example with a thickness of from 10 to 100 μ m).

If the substrate is sufficiently rigid, as is glass, it is also possible to adapt the screen-printing technology described above by etching shallow lines on the surface of the glass, parallel lines which will be filled with the screen-printing paste, giving a screen-printed network which is particularly discreet and is at the same time conducting.

Provision may advantageously be made for the two-dimensional network C of grid or fabric type, even in the event that it is self-supporting, to be embedded at the surface of the thermoplastic polymer film serving for lamination of the glazing. It may have been preembedded, like the conducting wires mentioned above. It may also be embedded in the film during the course of the lamination.

Rather than the screen printing of conducting wires, these wires may be laid down, for example as tungsten wires, on substrates which may have been previously provided with a conducting layer of doped metal oxide type, and retained at the periphery of the glazing by means of a suitable double-sided adhesive, which can also have a sealing function.

In one preferred embodiment of the first version according to the invention, the multicomponent electrode comprises at least one layer A and at least one layer B in electrical contact, at least one of these optionally being in contact with at least one layer D of dielectric material, and all of the superposed layers A, B and D preferably forming a stack

of layers with interference interaction. The stacks of layers found here may actually be rather similar to those used as low-emission/solar protection stacks for glazing in buildings or vehicles, stacks of the following outline type:

dielectric coating①/silver/dielectric coating②, with optionally a fine layer of protective metal at the Ag/dielectric interface. The dielectric coating may be a layer or superposed layers based on metal oxide (SnO₂, ZnO, TiO₂, SiO₂, Ta₂O₅, Nb₂O₅, etc.) or silicon nitride or oxynitride (SiON, Si₃N₄) or a mixture. Reference may be made, for example, to the stacks described in patents EP-0 611 213, EP-0 678 484 and EP-0 718 200, or else those used to equip glazing marketed in the "Planitherm" range by Saint-Gobain Vitrage. The application here is different and it is therefore necessary to adapt these stacks by replacing one or both of the abovementioned dielectric coatings ① and ② with one or more doped oxide conducting layers (of type A). The interposition of electrically insulating layers between the superposed conducting layers and the rest of the active layers in the system must be avoided. On the contrary, there is no reason not to add dielectric insulating layers to the conducting stacks of type (layer A/layer B) or (layer A/layer B/layer A) on the opposite side from the remainder of the functional layers, for example on the side facing the carrier substrate. The resultant stacks may be of the type:

substrate/dielectric D/layer B (Ag type)/layer A (ITO type)/remainder of the functional stack in the active system.

These layers D then fulfil an optical function and/or a function of anchoring type B layers to the substrate, and/or fulfil a function as a barrier to the migration of species coming from the substrate (for example alkalis coming from the glass). As mentioned above, the dielectric materials may be in the form of

oxide, oxycarbide or oxynitride of a metal or of silicon, or be based on silicon nitride.

Some examples of this type of electrode are ITO/Ag/ITO, Ag/ITO and dielectric/Ag/ITO with optional
5 interposition of thin layers of partially oxidized metal at the Ag/ITO interface, the second ITO layer protecting the silver layer while participating in the electrical conductivity of the entirety.

The multicomponent electrodes according to the
10 invention are provided with appropriate current leads of a type known in the art, in particular in the form of metal braids or shims.

As mentioned above, the invention is particularly applicable to an electrochromic system with at
15 least one carrier substrate and a stack of functional layers comprising at least, in succession, a first electrically conducting layer, an electrochemically active layer liable to reversible insertion of ions, such as H^+ , Li^+ or OH^- , of anodic or, respectively,
20 cathodic electrochromic material type, a layer of electrolyte, a second electrochemically active layer liable to reversible insertion of ions, such as H^+ , Li^+ or OH^- , of cathodic or, respectively, anodic electrochromic material type, and a second electrically
25 conducting layer, with at least one of the electrically conducting layers in the form of a layer A based on metal oxide(s) and being part of a multicomponent electrode E.

It is also applicable to any viologenic system
30 with at least one carrier substrate and a stack of functional layers comprising at least, in succession, a first electrically conducting layer, a film having viologenic properties in the form of a polymer, of a gel or of a suspension in a liquid medium, and a second
35 electrically conducting layer, with at least one of the two electrically conducting layers being of type A based on metal oxide(s) and being part of a multicomponent electrode E.

The invention thus relates to all the types of "active" system described in the preamble to the present application.

According to the invention it is advantageous
5 for the stack of functional layers to be arranged between two substrates, each of which may be rigid, of glass type or rigid polymer, such as polycarbonate or PMMA (methyl polymethacrylate), or semi-rigid or flexible of PET type (polyethylene terephthalate), all
10 of these being preferably transparent. They may also be absorbent or otherwise.

Another subject of the invention is glazing incorporating the device/active system described above, said device using as carrier substrate at least one of
15 the rigid constituent substrates of the glazing and/or at least one flexible substrate combined by lamination to one of the rigid constituent substrates of said glazing.

Another subject of the invention is the use of
20 the device and of the glazing described above for making glazing for buildings, in particular exterior glazing or internal partition glazing or door glazing or roof windows, glazing equipping internal partitions or windows or roof windows in means of transport of the
25 type comprising trains, aircraft, cars and boats, glazing for display screens of the type comprising computer or television screens, or touch screens, or spectacles, camera lenses or protection for solar panels.

Another subject of the invention is the use of
30 the device described above for making electrochemical energy-storing devices of battery or fuel-cell type, and the batteries and cells themselves. The reason for this is that it is of very particular interest for
35 battery applications to use the version of the invention consisting in the use of an electrode comprising a perforated metallic sheet or metallic grid. Since batteries are frequently produced on rather thin plastic substrates (of PET type, about 30 μm) if

the conducting layers are flexed they are in danger of losing their electrical continuity. A thicker metallic "grid" allows this continuity to be assured more effectively.

5 The advantageous characteristics and details of the invention are apparent from the description below of various non-limiting embodiments, with reference to the attached drawings, which show:

10 □ in **Figure 1**: viologenic glazing according to the invention

 □ in **Figure 2**: a first example of electrochromic glazing of "all-solid" type according to the invention

15 □ in **Figure 3**: the optical and electrical characteristics of the glazing according to Figure 2

 □ in **Figure 4**: a second example of electrochromic glazing of "all-solid" type according to the invention.

20 To make the figures easier to read they are highly diagrammatic and do not necessarily reflect the relative scale of the various elements which they show.

EXAMPLE 1

25 Figures 1a and 1b show a transverse section of a viologenic system using an "active" layer 3 of the type described in the abovementioned patent application EP-0 612 826 and based on polymer and arranged between two clear silica-soda-lime glass substrates 1 and 5 of 4 mm thickness. (Figure 1b is a view taking a section at right angles to Figure 1a).

30 The two substrates 1 and 5, each coated in advance with an SnO_2 :F layer 2 and 4 and laid down in a known manner by CVD have each then been provided with a network 6 and 7 of conducting strips, using a silver paste and the well known technique of screen printing.

35 The conducting strips have a width of 0.3 mm and are essentially parallel one to the other and separated one from the other by a distance of about 2 mm. A peripheral seal 8 ensures that the system does not leak.

There are therefore two multicomponent electrodes combining a screen-printed conducting network and a doped oxide layer. The layers of $\text{SnO}_2 : \text{F}$ may be replaced by a layer of ITO or $\text{SnO}_2 : \text{Sb}$, for example, and have a thickness of about 400 nm. It will be noted that the addition of a screen-printed network which raises the conductivity of the electrode can allow the conducting layers laid down to be of lower thickness while retaining the benefits of the invention, namely a diminution in the coloration front effect and a smaller switching time. If the thickness of the layer of $\text{SnO}_2 : \text{F}$ (or ITO) is reduced in this way there can be a significant reduction in the cost of the active glazing. The current leads are formed by screen printing perpendicular to the screen-printed conducting strips, parallel and at equal distances of 2 mm.

EXAMPLE 2

Figure 2 represents an embodiment of electrochromic glazing according to the invention: this is electrochromic glazing with a laminated structure and two panes of glass, in a configuration adapted, for example, for use as glazing in a car sun-roof: two clear panes of glass 21 and 22 are shown, with an electrochromic functional system 23 of "all-solid" type formed by the stack of functional layers which follows, and a sheet of polyurethane PU 24 (the sheet of PU may be replaced by a sheet of ethylene vinyl acetate EVA or of polyvinyl butyral PVB):

- ↪ a first 400 nm electrically conducting layer 25 of $\text{SnO}_2 : \text{F}$, laid down on the glass 22 using CVD,
- ↪ a first 40 nm layer 26 of anodic electrochromic material made of (hydrated) iridium oxide IrO_xH_y (which may be replaced by a layer of hydrated nickel oxide),
- ↪ a 100 nm layer 27 of tungsten oxide,
- ↪ a second 100 nm layer 28 of hydrated tantalum oxide,
- ↪ a second 370 nm layer 29 of cathodic electrochromic material based on tungsten oxide H_xWO_3 ,
- ↪ a second 50 nm layer 30 of ITO.

The entirety of glass pane 22/functional system 23 is then laminated to glass pane 21 via the sheet 24 of PU of thickness at least 1.24 mm which has been functionalized by laying down a network 31 of linear
5 metallic wires parallel one to the other. (It is also possible for this to be a sheet of EVA or of PVB, as stated above, for example with a thickness of the order of 0.76 mm).

The network is laid down in a known manner by
10 the process described in the abovementioned patents. As is known, the current leads are two shims arranged on the opposing margins of the sheet 24 of PU, applied with the aid of a soldering iron. They may also be metallic wire braids. The electrical contact between
15 these current leads (not shown) and the underlying electrically conducting layer is obtained by pressure, at the time of the lamination.

The glazing therefore uses a standard electrode on the glass pane 22, namely a monolayer of $\text{SnO}_2 : \text{F}$ (or
20 of ITO, for example) and a second electrode according to the invention combining an electrically conducting layer of ITO with a network of metallic wires. As in Example 1, this configuration allows the use of ITO layers alongside the PU film which are thinner than
25 those necessary in the absence of the conducting network 31. This network is formed, for example, by linear parallel wires, made of tungsten or of copper, optionally covered with graphite, of diameter $25 \mu\text{m}$ on average (for example between 10 and $50 \mu\text{m}$). Each line
30 is separated from the adjacent line by a distance of 2 mm (for example between 1 and 5 mm). This dimensioning is appropriate so that the network, although visible at very close quarters, is still very discreet and even invisible in the coloured state, an aesthetic
35 requirement which applies in the context of glazing for a car roof.

Figure 3 indicates the optical and electrical behaviour of glazing according to this example and measuring $35 \times 35 \text{ cm}^2$.

The graph 3 describes the optical appearance and the electrical behaviour of the glazing during switching. The X axis gives the time T expressed in seconds and the Y axis (on the left) gives the value
 5 for light transmission T_L expressed in % and (on the right) the current i in mA at the terminals of the glazing. The curve C1 gives the change in T_L at the edge of the glazing and the curve C2 gives the change in T_L at the centre of the glazing. It can be seen that
 10 the two curves are (almost) superposed, and this proves the absence or the virtual absence of coloration front. The curve C3 shows how the current i changes.

EXAMPLE 3

Figure 4 shows another version of "all-solid"
 15 electrochromic glazing according to the invention. As in Figure 2 and Example 2, two panes of glass 21 and 22 are seen, combined by lamination with the aid of a film 24 made of PU (or PVB: polyvinyl butyral), the layer 26 made of anodic electrochromic material, the
 20 layer 29 made of cathodic electrochromic material, these being separated by the layers 27 and 28, forming the electrolyte. By way of contrast, the electrode 25' arranged on the glass pane 22 is now formed by a stack of layers comprising a 34 nm layer 25(a) of SnO_2 on
 25 which there is a 10 nm layer 25(b) of silver, on which there is in turn a 50 nm layer 25(c) of ITO. This three-layer structure is obtained by cathodic sputtering assisted by a magnetic field, in a known manner. The layer 25(b) of silver is optionally provided with a
 30 thin layer 25(d) of metal intended to protect it during laying down of the layer 25(c) of ITO, when this latter is laid down by a reactive method in the presence of oxygen.

The resultant stack is rendered highly
 35 conducting by the presence of the layer of Ag, the light reflection from which is lowered with the help of the underlying layer of SnO_2 and of the layer of ITO lying thereupon, which serve as antireflective layers, by appropriate choice of their thickness. It will be

noted that it is necessary here for the layer 25(c) on the silver to be conducting in order to ensure that a voltage is applied to the remainder of the functional layers in the system, but this is not necessary for the layer 25(a) underneath the silver, which essentially has an optical function and which is an insulating dielectric. It is certainly possible to consider replacing this layer (totally or partially) by ITO or $\text{SnO}_2 : \text{F}$ in order to retain its optical function while at the same time again increasing the conductivity of the entirety of the layers 25(a), 25(b) and 25(c) in the multicomponent electrode.

The second electrode 30' is also a multilayer stack, for example one laid down by cathodic sputtering and composed of a first 50 nm layer 30(a) of ITO, of a second 10 nm layer 30(b) of silver and finally of a third 34 nm layer 30(c) of ITO. It is preferable here for the layer 30(a) and the layer 30(c) to be conducting, although they fulfil the same optical function with respect to the Ag layer 31(a) as the layers 25(a) and 25(c) with respect to the silver layer 25(b), since it is simpler to terminate the stack in a conducting layer to which the connecting elements can be affixed, these being here the metallic shims arranged on the sheet of polymer serving as a lamination intercalation.

Prior to the invention, a system of this type operated with a first 150 nm layer of ITO (adjacent to glass pane 21) and a second 300 nm layer of ITO (adjacent to PU 24). It can be seen, therefore, that the invention allows the use of much thinner layers of ITO or of $\text{SnO}_2 : \text{F}$, and this has a significant impact on the cost of the final glazing. The invention also allows the use of Ag layers with high electrical performance, but without the known disadvantages of these (very reflective appearance, a certain degree of fragility, etc.).

CLAIMS

1. Electrochemical device, in particular an electrically controllable system having variable optical and/or energy properties, including at least one carrier substrate provided with a stack of functional layers comprising at least one electrically conducting layer A based on metal oxide(s) and at least one electrochemically active layer F, **characterized in that** said layer A is part of a multicomponent electrode E combining with the layer A at least one higher-conductivity material B and/or at least one network C of conducting wires or of conducting strips.
2. Device according to Claim 1, **characterized in that** the material B is in the form of at least one layer combined with the layer A and in electrical contact therewith.
3. Device according to Claim 1, **characterized in that** the material B is incorporated in the layer A, in particular in the form of fibres or particles.
4. Device according to one of the preceding claims, **characterized in that** the layer(s) A is/are based on doped metal oxides chosen from at least one of the following doped oxides: doped tin oxide, in particular fluorine- or antimony-doped, doped zinc oxide, in particular aluminium-, tin- or fluorine-doped, and doped indium oxide, in particular tin-doped (ITO).
5. Device according to one of the preceding claims, **characterized in that** the material B is essentially metallic, in particular based on metals or on their alloys, for example Ag, Au, Cu, Al, or on alloys of Ag with another metal, in particular nickel or titanium.
6. Device according to one of the preceding claims, **characterized in that** said multicomponent electrode E is essentially transparent in the visible region.
7. Device according to one of the preceding claims, **characterized in that** said network C includes a

plurality of conducting strips, in particular essentially parallel each to the other and obtained by screen printing using a paste-like suspension of silver-type metal and a low-melting-point frit in an organic binder.

8. Device according to Claim 7, **characterized in that** the network C is screen-printed onto the glass-type carrier substrate, then covered with at least one electrically conducting layer A in order to form an electrode E, or is laid down on the electrically conducting layer A covering the carrier substrate.

9. Device according to one of Claims 1 to 5, **characterized in that** the network C includes a plurality of conducting wires in the form of essentially metallic wires surface-laid on a sheet based on thermoplastic polymer.

10. Device according to one of Claims 1 to 5, **characterized in that** the network C is based on a fabric, a net or a metallic nonwoven, in particular a network obtained using metallic wires of diameter between 10 and 100 μm , and is preferably a network surface-laid on a sheet based on thermoplastic polymer.

11. Device according to one of Claims 1 to 5, **characterized in that** the network C is obtained by etching or perforating a metallic layer or a metallic sheet.

12. Device according to one of the preceding claims, **characterized in that** the multicomponent electrode E comprises at least one layer A and at least one layer B in electrical contact, at least one of these layers optionally being in contact with at least one layer D of dielectric material, and all of the layers A, B and D preferably forming a stack of layers with interference interaction.

13. Device according to Claim 12, **characterized in that** the layer(s) D have an optical function and/or a function of anchoring the other layers B to the carrier substrate and/or a function as a barrier to the migration of alkaline species coming from the glass, in

particular in the form of oxide, oxycarbide or oxynitride of a metal or of silicon, or silicon nitride.

14. Device according to one of the preceding
5 claims, **characterized in that** the multicomponent electrode(s) E comprise(s) the sequence ITO/Ag/ITO or Ag/ITO with optional interposition of thin layers of partially oxidized metal at the Ag/ITO interface.

15. Device according to one of the preceding
10 claims, **characterized in that** the multicomponent electrode(s) E is/are provided with current leads, in particular in the form of metal braids or shims.

16. Device according to one of the preceding
15 claims, **characterized in that** it is an electrochromic system, in particular an "all-solid" or "all-polymer" electrochromic system, with at least one carrier substrate and a stack of functional layers comprising at least, in succession, a first electrically conducting layer, an electrochemically active layer liable to
20 reversible insertion of ions, such as H^+ , Li^+ or OH^- , of anodic-colouring or, respectively, cathodic-colouring electrochromic material type, a layer of electrolyte, a second electrochemically active layer liable to reversible insertion of ions, such as H^+ , Li^+ or OH^- , of
25 cathodic-colouring or, respectively, anodic-colouring electrochromic material type, and a second electrically conducting layer, with at least one of the two electrically conducting layers in the form of a layer A based on metal oxide(s) and being part of a multi-
30 component electrode E.

17. Device according to one of Claims 1 to 15, **characterized in that** it is a viologenic system with at least one carrier substrate and a stack of functional layers comprising at least, in succession, a first
35 electrically conducting layer, a film having viologenic properties in the form of a polymer, of a gel or of a suspension in a liquid medium, and a second electrically conducting layer, with at least one of the two electrically conducting layers being of type A

based on metal oxide(s) and being part of a multi-component electrode E.

18. Device according to one of the preceding claims, **characterized in that** the stack of functional
5 layers is arranged between two substrates, each of which may be rigid, of glass type or rigid polymer, such as polycarbonate or PMMA, or semi-rigid or flexible of PET type, all of these being preferably transparent or absorbent or otherwise.
19. Glazing **characterized in that** it incorporates the device according to one of the preceding claims, said device using as carrier substrate at least one of the rigid constituent substrates of the glazing and/or
15 at least one flexible substrate combined by lamination with one of the rigid constituent substrates of said glazing.
20. Use of the device according to one of Claims 1 to 18 or of the glazing according to Claim 19 for making glazing for buildings, in particular exterior
20 glazing or internal partition glazing or door glazing or roof windows, glazing equipping internal partitions or windows or roof windows in means of transport of the type comprising trains, aircraft, cars and boats, glazing for display screens of the type comprising
25 computer or television screens, or touch screens, or for making spectacles or camera lenses or protection for solar panels.
21. Use of the device according to one of Claims 1 to 18 for making electrochemical energy-storing devices
30 of battery or fuel-cell type.



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(54) Title: ELECTROCHEMICAL DEVICE, SUCH AS AN ELECTRICALLY CONTROLLED SYSTEM WITH VARIABLE OPTICAL AND/OR ENERGY PROPERTIES

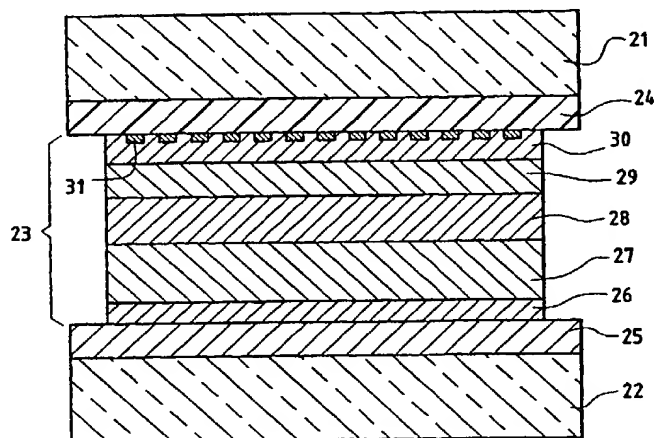
(54) Titre: DISPOSITIF ELECTROCHIMIQUE, DU TYPE SYSTEME ELECTROCOMMANDABLE A PROPRIETES OPTIQUES ET/OU ENERGETIQUES VARIABLES

(57) Abstract

The invention concerns an electrochemical device, in particular an electrically controlled system with variable optical and/or energy properties, comprising at least a substrate provided with a stack of functional layers including at least one electroconductive layer A based on metal oxide(s) and at least an electrochemically active layer F. Said layer A forms part of a multicomponent electrode E associating with the layer A at least a material B more conductive than the former and/or at least a network C of conductor wires or conduction bands. The invention also concerns its uses, in particular in glazing.

(57) Abrégé

L'invention a pour objet un dispositif électrochimique, notamment système électrocommandable à propriétés optiques et/ou énergétiques variables, comportant au moins un substrat porteur muni d'un empilement de couches fonctionnelles dont au moins une couche A électroconductrice à base d'oxyde(s) métallique(s) et au moins une couche F électrochimiquement active. Ladite couche A fait partie d'une électrode multi-composante E associant à la couche A au moins un matériau B plus conducteur qu'elle et/ou au moins un réseau C de fils conducteurs ou de bandes conductrices. L'invention a également pour objet ses applications, notamment dans des vitrages.



- 25 -

PATENT

**ELECTROCHEMICAL DEVICE OF ELECTRICALLY CONTROLLABLE
SYSTEM TYPE HAVING VARIABLE OPTICAL AND/OR ENERGY
PROPERTIES**

The subject of the invention is an electrochemical device, in particular an electrically controllable system having variable optical and/or energy properties, including at least one carrier substrate provided with a stack of functional layers comprising at least one electrically conducting layer A based on metal oxide(s) and at least one electrochemically active layer F. Said layer A is part of a multicomponent electrode E combining with the layer A at least one higher-conductivity material B and/or at least one network C of conducting wires or conducting strips.

Another subject of the invention is the applications of the device, in particular in glazing.

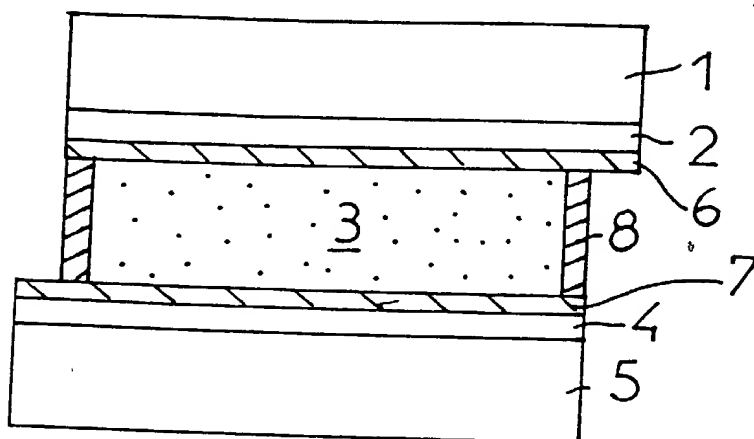


FIG-1a

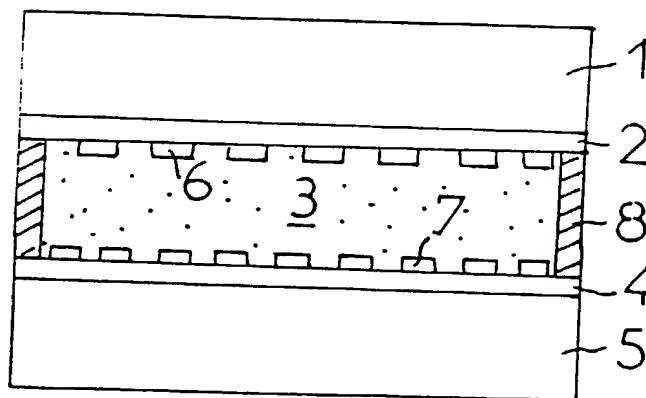


FIG-1b

FIG-1

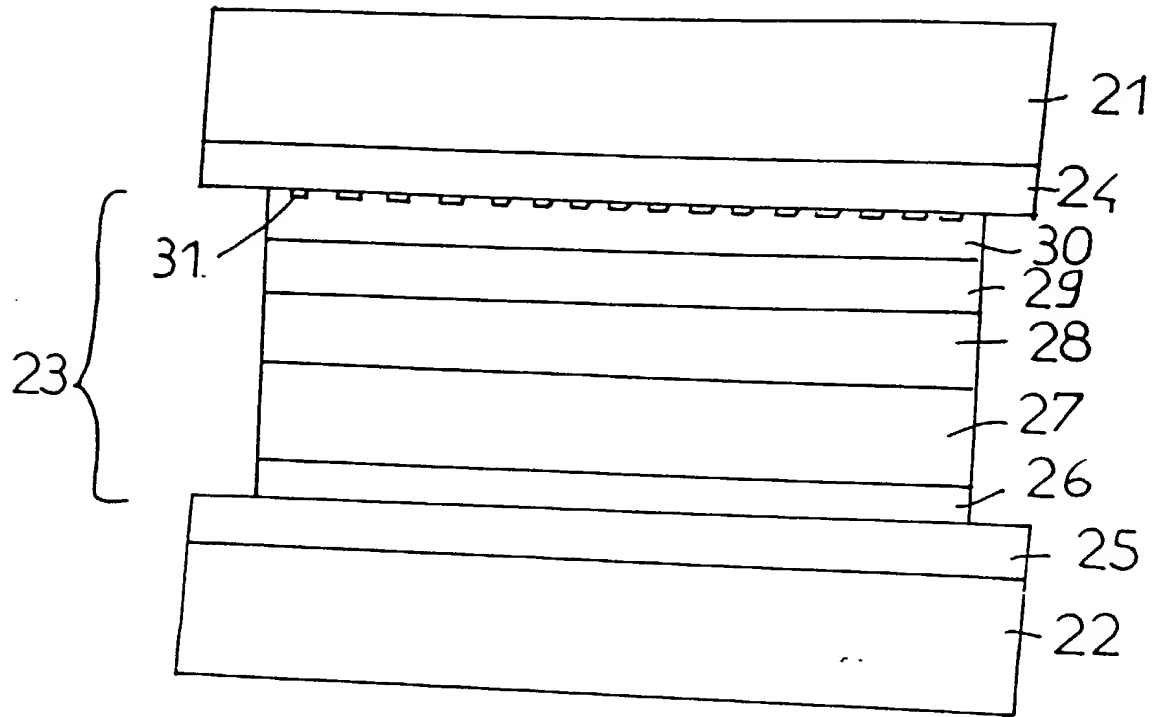


FIG-2

09/700684

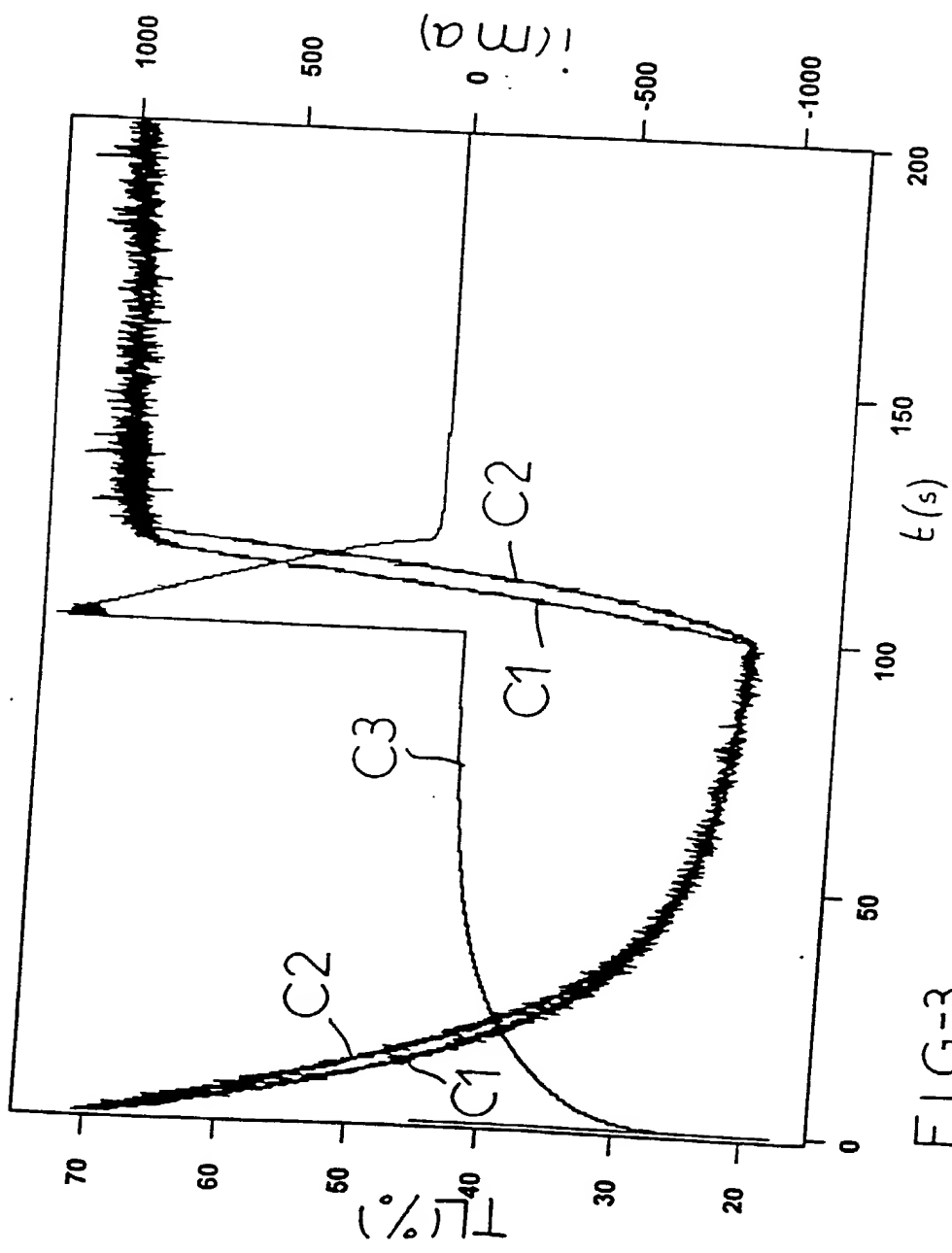
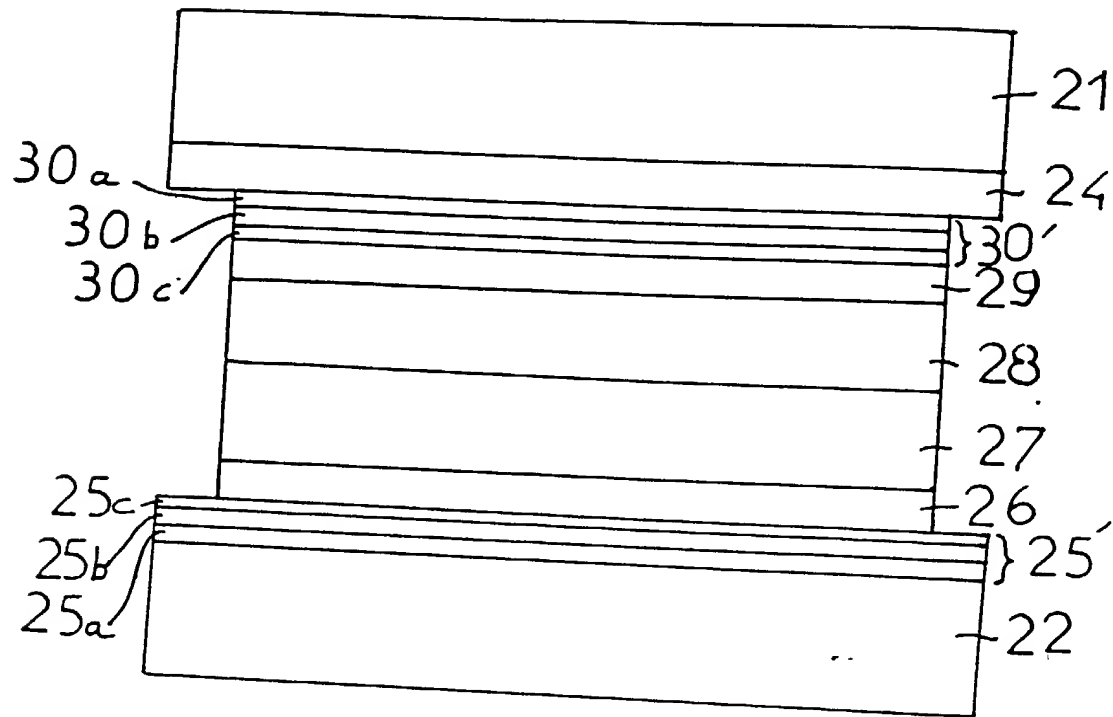


FIG-3

FIG-4

Declaration and Power of Attorney for Patent Application

Déclaration et Pouvoirs pour Demande de Brevet

French Language Declaration

En tant l'inventeur nommé ci-après je déclare par le présent acte que:

Mon domicile, mon adresse postale et ma nationalité sont ceux figurant ci-dessous à côté de mon nom.

Je crois être le premier inventeur original et unique (si un seul nom est mentionné ci-dessous), ou l'un des premiers co-inventeurs originaux (si plusieurs noms sont mentionnés ci-dessous) de l'objet revendiqué pour lequel une demande de brevet a été déposée concernant l'invention intitulée

et dont la description est fournie ci-joint à moins

☐ ci-joint

☐ a été déposée le _____

sous le numéro de demande des Etats-Unis ou le numéro de demande international PCT

_____ et modifiée le

_____ (le cas échéant).

Je déclare par le présent acte avoir passé en revue et compris le contenu de la description ci-dessus, revendications comprises, telles que modifiées par toute modification dont il aura été fait référence ci-dessus.

Je reconnais devoir divulguer toute information pertinente à la brevetabilité, comme défini dans le Titre 37, § 1.56 du Code fédéral des réglementations.

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled

ELECTROCHEMICAL DEVICE, SUCH AS AN

ELECTRICALLY CONTROLLED SYSTEM WITH

VARIABLE OPTICAL AND/OR ENERGY PROPERTIES
(as amended)

the specification of which:

☐ is attached hereto.

☒ was filed on November 17, 2000

as United States Application Number or PCT International Application Number

_____ and was amended on

_____ (if applicable).

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56.

French Language Declaration

Je revendique par le présent acte avoir la priorité étrangère, en vertu du Titre 35, § 119(a)-(d) ou § 365(b) du Code des Etats-Unis, sur toute demande étrangère de brevet ou certificat d'inventeur ou, en vertu du Titre 35, § 365(e) du même Code, sur toute demande internationale PCT désignant au moins un pays autre que les Etats-Unis et figurant ci-dessous et, en cochant la case, j'ai aussi indiqué ci-dessous toute demande étrangère de brevet, tout certificat d'inventeur ou toute demande internationale PCT ayant une date de dépôt précédant celle de la demande à propos de laquelle une priorité est revendiquée.

I hereby claim foreign priority under Title 35, United States Code, § 119(a)-(d) or § 365(b) of any foreign application(s) for patent or inventor's certificate, or § 365(a) of any PCT International application which designated at least one country other than the United States, listed below, and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or PCT International application having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)
Demande(s) de brevet anterieure(s) dans un autre pays.

Priority claimed
Droit de priorité
revendiqué

99/03420	FRANCE
(Number)	(Country)
(Numéro)	(Pays)

19-March 1999
(Day/Month/Year Filed)
(Jour/Mois/Anné de dépôt)

<input checked="" type="checkbox"/>	<input type="checkbox"/>
Yes	No
Qui	Non

(Number) (Numéro)	(Country) (Pays)
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(Day/Month/Year Filed)
(Jour/Mois/Anné de dépôt)

<input type="checkbox"/>	<input type="checkbox"/>
Yes	No
Oui	Non

Je revendique par le présent acte tout bénéfice, en vertu du Titre 35, § 119(e) du Code des Etats-Unis, de toute demande de brevet provisoire effectuée aux Etats-Unis et figurant ci-dessous.

I hereby claim the benefit under Title 35, United States Code, § 119(e) of any United States provisional application(s) listed below.

(Application No.)
(N° de demande)

(Filing Date)
(Date de dépôt)

(Application No.)
(N° de demande)

(Filing Date)
(Date de dépôt)

Je revendique par le présent acte tout bénéfice, en vertu du Titre 35, § 120 du Code des Etats-Unis, de toute demande de brevet effectuée aux Etats-Unis, ou en vertu du Titre 35, § 365(c) du même Code, de toute demande internationale PCT désignant les Etats-Unis et figurant ci-dessous et, dans la mesure où l'objet de chacune des revendications de cette demande de brevet n'est pas divulgué dans la demande antérieure américaine ou internationale PCT, en vertu des dispositions du premier paragraphe du Titre 35, § 112 du Code des Etats-Unis, je reconnais devoir divulguer toute information pertinente à la brevetabilité, comme défini dans le Titre 37, § 1.56 du Code fédéral des réglementations, dont j'ai pu disposer entre la date de dépôt de la demande antérieure et la date de dépôt de la demande nationale ou internationale PCT de la présente demande:

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s), or § 365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International application in the manner provided by the first paragraph of Title 35, United States Code, § 112, I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56 which became available between the filing date of the prior application and the national or PCT International filing date of this application.

PCT/FR00/00675

17 March 2000

(Application No.)
(N° de demande)

(Filing Date)
(Date de dépôt)

(Status) (patented, pending, abandoned)
(Statut) (breveté, en cours d'examen, abandonné)

(Application No.)
(N° de demande)

(Filing Date)
(Date de dépôt)

(Status) (patented, pending, abandoned)
(Statut) (breveté, en cours d'examen, abandonné)

Je déclare par le présent acte que toute déclaration ci-incluse est, à ma connaissance, véridique et que toute déclaration formulée à partir de renseignements qui de suppositions est tenue pour véridique; et de plus, que toutes ces déclarations ont été formulées en sachant que toute fausse déclaration volontaire ou son équivalent est passible d'une amende ou d'une incarcération, ou des deux, en vertu de la Section 1001 du Titre 18 du Code des Etats-Unis, et que ces telles déclarations volontairement fausses risquent de compromettre la validité de la demande de brevet ou du brevet délivré à partir de celle-ci.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

French Language Declaration

POUVOIRS: En tant que l'inventeur ci-dessus, je désigne par la présente l'(les) avocat(s) et/ou agent(s) suivant(s) pour qu'ils poursuive(nt) la procédure de cette demande de brevet et traite(nt) toute affaire s'y rapportant avec l'Office des brevets et des marques: (*mentionner le nom et le numéro d'enregistrement*).

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith: (*list name and registration number*)

Norman F. Oblon, Reg. No. 24,818; Marvin J. Spivak, Reg. No. 24,913; C. Irvin McClelland, Reg. No. 31,124; Gregory J. Maier, Reg. No. 25,599; Arthur I. Neustadt, Reg. No. 24,854; Richard D. Kelly, Reg. No. 27,757; James D. Hamilton, Reg. No. 28,421; Eckhard H. Kuesters, Reg. No. 28,970; Robert T. Pous, Reg. No. 21,099; Charles L. Gholz, Reg. No. 26,395; William E. Beaumont, Reg. No. 30,996; Jean-Paul Lavalleye, Reg. No. 31,451; Stephen G. Baxter, Reg. No. 32,884; Richard L. Treanor, Reg. No. 36,379; Steven P. Weihrouch, Reg. No. 32,829; John T. Goolkasian, Reg. No. 26,142; Richard L. Chinn, Reg. No. 34,305; Steven E. Lipman, Reg. No. 30,011; Carl E. Schlier, Reg. No. 34,426; James J. Kulbaski, Reg. No. 34,448; Richard A. Neifeld, Reg. No. 35,299; J. Derek Mason, Reg. No. 35,270; Surinder Sachar, Reg. No. 34,423; Christina M. Gadlano, Reg. No. 37,628; Jeffrey B. McIntyre, Reg. No. 36,867; William T. Enos, Reg. No. 33,128; Michael E. McCabe, Jr., Reg. No. 37,182; Bradley D. Lytle, Reg. No. 40,073; and Michael R. Casey, Reg. No. 40,294 with full powers of substitution and revocation.

Addresser toute correspondance à:

Send Correspondence to:

OBLON, SPIVAK, MCCLELLAND, MAIER & NEUSTADT, P.C.
FOURTH FLOOR
1755 JEFFERSON CAVIS HIGHWAY
ARLINGTON, VIRGINIA 22202 U.S.A.

Adresser tout appel téléphonique à:
(nom et numéro de téléphone)

Direct Telephone calls to: (name and telephone number)

(703) 413-3070

Nom complet de l'unique ou premier inventeur		Full name of sole or first inventor	
Signature de l'inventeur		Inventor's signature	
Date		Date	
Domicile		Residence	
Nationalité		Citizenship	
Adresse Postale		Post Office Address	
Nom complet du second co-inventeur, le cas échéant		Full name of second joint inventor, if any	
Signature de l'inventeur		Second inventor's signature	
Date		Date	
Domicile		Residence	
Nationalité		Citizenship	
Adresse Postale		Post Office Address	

(Fournir les mêmes renseignements et la signature de tout co-inventeur supplémentaire.)

(Supply similar information and signature for third and subsequent joint inventors.)

French Language Declaration

Nom complet du troisième co-inventeur, le cas échéant Signature de l'inventeur Domicile Nationalité Adresse Postale	Full name of third joint inventor, if any Third Inventor's signature Residence Citizenship Post Office Address
300 Date	Jean-Christophe GIRON Date 03.01.01
	36, rue Traversière E-75012 Paris, France FRA
	France
	same as above

Nom complet du quatrième co-inventeur, le cas échéant Signature de l'inventeur Domicile Nationalité Adresse Postale	Full name of fourth joint inventor, if any Fourth Inventor's signature Residence Citizenship Post Office Address
Date	Date

Nom complet du cinquième co-inventeur, le cas échéant Signature de l'inventeur Domicile Nationalité Adresse Postale	Full name of fifth joint inventor, if any Fifth Inventor's signature Residence Citizenship Post Office Address
Date	Date

Nom complet du sixième co-inventeur, le cas échéant Signature de l'inventeur Domicile Nationalité Adresse Postale	Full name of sixth joint inventor, if any Sixth Inventor's signature Residence Citizenship Post Office Address
Date	Date

(Fournir les mêmes renseignements et la signature de tout co-inventeur supplémentaire.)

(Supply similar information and signature for third and subsequent joint inventors.)